

# TiO<sub>2</sub>: Au (Titanium Dioxide Conjugated Gold) Crystals as the Biomimetic Oxidase for Rapid and Sensitive Oxalate Detection Combining with Bionic E-Eye

K. Gayathiridevi<sup>1,2</sup>, G. Pasupathi<sup>1,3</sup>

<sup>1</sup>Research and Development Center, Bharathidasan University, Thiruchirappalli, 620 025, India

<sup>2</sup>Department of Physics, Bharathidasan University Constituent College for Women, Orathanadu -614 625, India

<sup>3</sup>Department of Physics, A.V.V.M. Sripushpam College (Auto), Poondi -613 503, India

**Abstract:** *TiO<sub>2</sub>:Au (Titanium dioxide conjugated Gold) crystal is a new materials was grown from aqueous solution. Solubility of the crystal has a positive temperature coefficient facilitating growth by slow cooling. Transmission electron microscopy and scanning electron microscopy are used to study the morphological studies of the crystals. Second harmonic generation (SHG) conversion efficiency of up to 2.56 times that of a phase matched conjugate crystal was achieved when the (TiO<sub>2</sub>:Au) crystal was rotated about c axis, by 9.1 in the clockwise direction. Furthermore, the method was tested by the artificial urine samples, indicating its greater performances for monitoring and diagnosis of urolithiasis in point-of-care applications.*

**Keywords:** oxalate, TiO<sub>2</sub>-Au conjugate crystals, bionic E-eye and urolithiasis

## 1. Introduction

Titanium dioxide with Gold conjugate is excellent combination of physical and chemical properties is potential but less explored candidates for optical properties. They are a class of organic materials that crystallize in symmetric space groups. Due to multidirectional hydrogen bonding, the crystals can be grown to large sizes from solution. They also possess higher hardness and melting temperature than van der Waals bonded organic crystals [1]. Urinary tract infection for the human is due to the Urolithiasis gives the symptoms such as shivering, fatigue, dysuria and renal pain. Depending on where a stone is located, it may be called a kidney stone, ureteral stone, or bladder stone. The process of stone formation is called urolithiasis [2]. Therefore, to monitor this disease for the prevention and treatment, measuring the oxalate in the urine is of great significance. Here, a rapid and sensitive colorimetric method was developed based on Titanium dioxide with Gold conjugate (TDG) for oxalate detection. Titanium dioxide with Gold conjugate (TDG) acted as an efficient biomimetic oxidase to catalyze the reaction. Yellow TDG can be oxidized to blue oxide and oxalate can selectively inhibit this reaction by consuming and reacting with TDG, thus achieving the quantitative detection of oxalate. Moreover, a home-made bionic electronic-eye (E-eye) system was developed as a portable in-situ detection platform to efficiently measure the oxalate concentrations in 10 s by direct photographing [3]. By optimizing experimental conditions, this method shows a wide linear range (7.1  $\mu\text{M}$  to 300  $\mu\text{M}$ ) and a low detection limit (0.89  $\mu\text{M}$ ) for oxalate detection. Besides, this method exhibits high selectivity even with 80-fold interfering chemicals [4].

The probability of stone formations varies with different countries of atmosphere, food and life syle. It varies from 1% to 10% in Asia, 5% to 9% in Europe, and 13% in North America [5]. It is formed due to the deposition of crystals in the kidney it consists of calcium phosphate, calcium oxalate and calcium carbonate, magnesium ammonium phosphate and cysteine [6]. It is worth stressing crystals in that 85% urolithiasis mainly contains calcium oxalate [7]. Normal oxalate levels in healthy people are 100-300  $\mu\text{M}$  in urine [8] and 0.8-2.50  $\mu\text{M}$  in plasma [9]. The excess concentration of oxalate in urine (>300  $\mu\text{M}$ ) indicates the high suspicion of urolithiasis. Therefore, an efficient method for measuring oxalate is of great necessity to prevent and diagnose urolithiasis in the early stage [10].

## 2. Experimental

The presence of Titanium in well known optical crystals, Titanium carbide (TiC) 99%, Auric chloride (AuCl<sub>3</sub>) 0.5%, and Glycine 0.5% was added. Also such a metal organic compound is expected to combine superior physical properties of their counterparts. This combination was diluted in the 100 mL solution and it was grown as a single crystal within 21 days. This crystals was ground in the mortar and then annealed to 200° C for 1 hour and then characterized by using UV-Vis absorbance and Photoluminescence studies florescence and it was used to detect oxalate detection combining with a highly portable bionic E-eye system. NaCl, KCl, glucose, Na<sub>2</sub>HPO<sub>4</sub>.12H<sub>2</sub>O, NH<sub>4</sub>Cl, AuCl<sub>3</sub>, TiC and Sodium citrate were purchased from Sigma Aldrich Chemical Reagent (Ltd). The kit for Oxalate detection was analyzed from Life Tech research Centre, Chennai. Bionic E-eye developed in Life Tech Research Centre, Chennai, laboratory.

### 3. Results and Discussion

The pH and the depth of the color is positively correlated with the concentration of oxalate solution and artificial urine solution. The absorbance was measured at 325 nm using a plate reader, and the oxalate content in the sample was calculated based on the standard curve. The fluorescence peaks observed for this oxalate solution is nearly 450 nm. The absorbance was measured 360 nm and fluorescence peak was nearly 440 nm also correlated with this same concentration. Absorbance and Fluorescence spectra values were compared with the results of Figures 1 and 2 for oxalate and artificial urine solution [11].

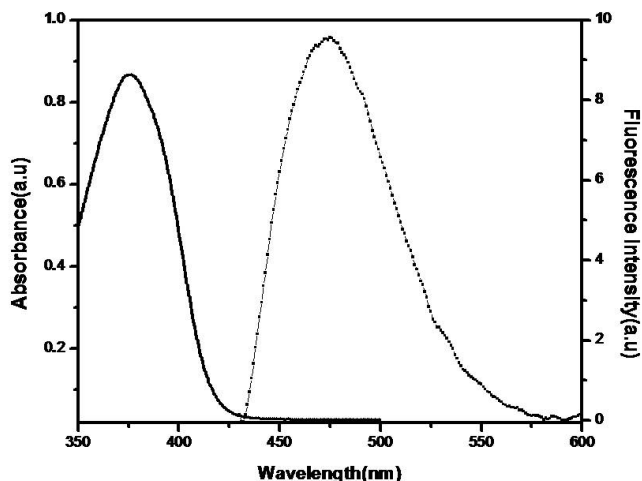


Figure 1: Absorbance and Fluorescence spectra of  $\text{TiO}_2\text{-Au}$  crystal in oxalate solution

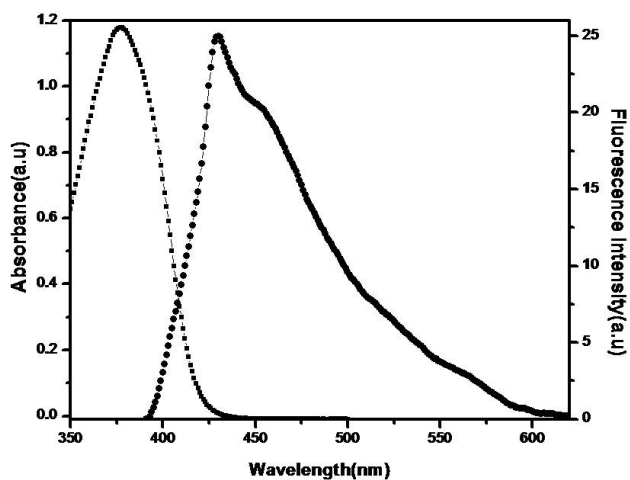


Figure 2: Absorbance and Fluorescence spectra of  $\text{TiO}_2\text{-Au}$  crystal in artificial Urine Solution

Compared to other materials/methods for oxalate detection (Table 1),  $\text{TiO}_2\text{-Au}$  single crystals demonstrate the lowest LOD and the shortest time cost. In plate reader, 96 wells are measured sequentially with a relatively long time cost. Home-made bionic E-eye can measure 96 wells which significantly shorten the plate reading time and ensure that all wells are detected simultaneously. Furthermore, the bionic E-eye with small size features and high portability, which is more suitable for on-site detection and point of care application [12].

Table 1:  $\text{TiO}_2\text{-Au}$  (average time required for one sample is 30 min)

Material	Artificial Urine Solution (LOD/ $\mu\text{M}$ )	Oxalate Solution (LOD/ $\mu\text{M}$ )
$(\text{TiO}_2\text{-Au})$ crystals (Plate reader)	31.80	26.23

### 4. Conclusion

In this work, we developed a simple, rapid and sensitive colorimetric method using based on TDG crystals for oxalate detection combining with a highly portable bionic E-eye system. Experimental conditions including the amount of TDG, interaction time of color reaction time, and pH have been optimized to ensure the ideal performance. The method shows a low LOD of  $0.89 \mu\text{M}$  and a wide detection range from  $6.9 \mu\text{M}$  to  $300 \mu\text{M}$  with very good repeatability, long-term stability and high selectivity. Artificial urine samples were measured with good recovery to validate the performance for real sample analysis. Compared to other methods, the bionic E-eye based method shows the lowest LOD and the shortest time cost with high portability.

### References

- [1] Edwards, J.K., Solsona, B.E., Landon, P., Carley, A.F., Herzing, A., Kiely, C.J. and Hutchings, G.J., 2005. Direct synthesis of hydrogen peroxide from  $\text{H}_2$  and  $\text{O}_2$  using  $\text{TiO}_2$ -supported Au-Pd catalysts. *Journal of Catalysis*, 236(1), pp.69-79.
- [2] Grunwaldt, J.D., Maciejewski, M., Becker, O.S., Fabrizioli, P. and Baiker, A., 1999. Comparative study of Au/ $\text{TiO}_2$  and Au/ $\text{ZrO}_2$  catalysts for low-temperature CO oxidation. *Journal of Catalysis*, 186(2), pp.458-469.
- [3] Kaviyarasu, K., A. Mariappan, K. Neyvasagam, A. Ayeshamariam, P. Pandi, R. Rajeshwara Palanichamy, C. Gopinathan, Genene T. Mola, and M. Maaza. "Photocatalytic performance and antimicrobial activities of HAp- $\text{TiO}_2$  nanocomposite thin films by sol-gel method." *Surfaces and Interfaces* 6 (2017): 247-255.
- [4] Kaviyarasu, K., Geetha, N., Kanimozhi, K., Magdalane, C. M., Sivaranjani, S., Ayeshamariam, A., ... & Maaza, M. (2017). In vitro cytotoxicity effect and antibacterial performance of human lung epithelial cells A549 activity of zinc oxide doped  $\text{TiO}_2$  nanocrystals: investigation of bio-medical application by chemical method. *Materials Science and Engineering: C*, 74, 325-333.
- [5] Mitra, P., Pal, D.K., Das, M., 2018. Does quality of drinking water matter in kidney stone disease: A study in West Bengal, India. *Investig Clin Urol* 59(3), 158-165.
- [6] Parmar, M.S., 2004. Kidney stones. *Bmj* 328(7453), 1420-1424.
- [7] Coe, F.L., Worcester, E.M., Evan, A.P., 2016. Idiopathic hypercalciuria and formation of calcium renal stones. *Nature reviews. Nephrology* 12(9), 519-533.
- [8] Pundir, C.S., Kuchhal, N.K., Thakur, M., Satyapal, 1998. Determination of plasma oxalate with chloride ion insensitive oxalate oxidase. *Indian journal of biochemistry & biophysics* 35(2), 120-122.
- [9] Pundir, C.S., Chauhan, N., Rajneesh, Verma, M., Ravi, 2011. A novel amperometric biosensor for oxalate

determination using multi-walled carbon nanotube-gold nanoparticle composite Sensor Actuat B-Chem 155(2), 796-803.

- [10] Armelao, L., Barreca, D., Bottaro, G., Gasparotto, A., Maccato, C., Maragno, C., Tondello, E., Štangar, U.L., Bergant, M. and Mahne, D., 2007. Photocatalytic and antibacterial activity of  $\text{TiO}_2$  and  $\text{Au/TiO}_2$  nanosystems. *Nanotechnology*, 18(37), p.375709.
- [11] Zhu, S., Liang, S., Gu, Q., Xie, L., Wang, J., Ding, Z. and Liu, P., 2012. Effect of Au supported  $\text{TiO}_2$  with dominant exposed  $\{0\ 0\ 1\}$  facets on the visible-light photocatalytic activity. *Applied Catalysis B: Environmental*, 119, pp.146-155.
- [12] Liu, G., Sun, C., Yang, H.G., Smith, S.C., Wang, L., Lu, G.Q.M. and Cheng, H.M., 2010. Nanosized anatase  $\text{TiO}_2$  single crystals for enhanced photocatalytic activity. *Chemical Communications*, 46(5), pp.755-757.